

Coulomb's Law :

Charles Augustin de Coulomb, who investigated the repulsion between small balls that he charged by rubbing process. To measure the force between the balls, he used a delicate torsion balance similar to the torsion balance later used by Henry Cavendish to measure gravitational forces. His experimental results are summarized in Coulomb's law:

- " The magnitude of the electric force that a particle ~~exerts~~ exerts on another particle is
- (i) directly proportional to the product of their charges;
 - (ii) inversely proportional to the square of the distance between them.
 - (iii) The direction of force is along the line joining the particles.

Mathematically, the electric force F that a particle of charge q_1 exerts on particle of charge q_2 at a distance r is given by

$$F = k \frac{q_1 q_2}{r^2}$$

where k is proportionality constant.

Note 1

The electric force that the particle of charge q_1 exerts on the particle of charge q_2 has the same magnitude as force exerted by q_2 on q_1 .

Note 2

(i) Like charges repel each other.

(ii) Unlike charges attract each other.

S.I unit value of k is constant

$$k = 9 \times 10^9 \frac{\text{N-m}^2}{\text{C}^2} \approx 9 \times 10^9 \frac{\text{N-m}^2}{\text{C}^2} \left(\frac{\text{m}}{\text{F}} \right)$$

Mathematically

$$k = \frac{1}{4\pi\epsilon_0}$$

The quantity ϵ_0 ("epsilon nought") is called the electric constant or the permittivity constant.

* the value of $\epsilon_0 = 8.85 \times 10^{-12} \frac{\text{C}^2}{\text{N-m}^2} (\text{F/m})$

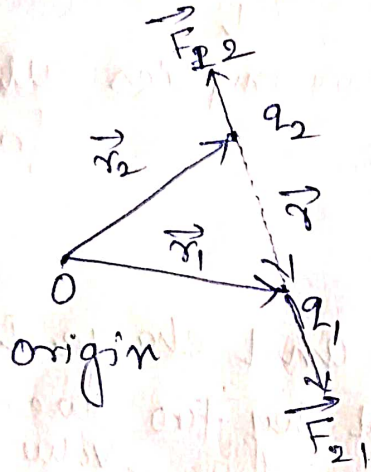
In terms of permittivity constant - Coulomb's law can be written as

$$F = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r^2}$$

Coulomb's law in vector notation.

$$\vec{F} = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r^2} \hat{r}$$

Let the location of charge q_1 is \vec{r}_1 and q_2 is \vec{r}_2



\vec{F}_{12} is force on q_2 due to q_1

$$\vec{r} = \vec{r}_2 - \vec{r}_1$$

$$\therefore \vec{F}_{12} = k \frac{q_1 q_2}{|\vec{r}_2 - \vec{r}_1|^2} \hat{r} \quad \leftarrow \text{unit vector}$$

$$\vec{F}_{12} = k \frac{q_1 q_2}{r^2} \hat{r}$$

$$\vec{F}_{12} = k \frac{q_1 q_2}{|\vec{r}_2 - \vec{r}_1|^2} \frac{(\vec{r}_2 - \vec{r}_1)}{|\vec{r}_2 - \vec{r}_1|}$$

$$= k \frac{q_1 q_2}{|\vec{r}_2 - \vec{r}_1|^3} (\vec{r}_2 - \vec{r}_1)$$

\vec{F}_{21} is force on q_1 due to q_2

$$\vec{F}_{21} = |\vec{F}_{12}| (-\hat{r})$$

$$\vec{F}_{21} = -\vec{F}_{12}$$

Note :

Sign of charges must be taken in account so that

(i) For like charges $q_1 q_2 > 0$ implies repulsive force

(ii) For unlike charges $q_1 q_2 < 0$

↓
implies attractive force

Problem 1 :

Two point charges of equal mass m and charge q are suspended at a common point by two threads of negligible mass and length l . Find the value of inclination of mass α ? If the angle of inclination of two masses are small and same.

Solution :

Consider system of charges as shown in figures

Where F_e is Coulomb force

T is the tension in each thread

mg is weight of each charge

$$T \sin \alpha = F_e \quad \text{--- (1)}$$

$$T \cos \alpha = mg \quad \text{--- (2)}$$

On dividing eqn (1) and eqn (2)

we get $\tan \alpha = \frac{F_e}{mg}$

$$\tan \alpha = \frac{1}{4\pi\epsilon_0} \frac{q^2}{r^2} \frac{1}{mg}$$

where r is distance between two particles

$$r = 2l \sin \alpha$$

for small α

$$\sin \alpha \approx \tan \alpha \approx \alpha$$

$$\therefore \alpha = \frac{1}{4\pi\epsilon_0} \frac{q^2}{4l^2 \alpha^2} \frac{1}{mg}$$

$$\Rightarrow \alpha^3 = \frac{q^2}{16\pi\epsilon_0 mg l^2}$$

